

W - 2022

**DR. BABASAHEB AMBEDKAR TECHNOLOGICAL UNIVERSITY,
LONERE-RAIGAD-402103**

End Semester Examination – Winter 2022

Course: B. Pharmacy

Subject Name: Human Anatomy & Physiology-I

Date: 04/05/2023

Max Marks:75

Semester: I

Subject Code: BP101T

Duration: 3 Hr.

Instructions:

1. All questions are compulsory
2. Draw diagrams / figures wherever necessary
3. Figures to right indicate full marks

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- Q. 1. Objective Type Questions (Answer all the questions) (10 x 2) = 20**
- i) Define human anatomy and physiology.
 - ii) Enlist different systems of human body.
 - iii) Define homeostasis.
 - iv) Define tissue and enlist different types of tissue.
 - v) Give the functions of skeletal system.
 - vi) How many bones are present in Appendicular skeleton?
 - vii) Difference between endocrine and exocrine gland.
 - viii) Define mitosis and meiosis.
 - ix) What do you mean by cardiac cycle?
 - x) Define cardiac output and cardiac reserve.
- Q. 2. Long Answers (Answer 2 out of 3) (2 x 10) = 20**
- i) Define integumentary system, explain in detail skin with the help of neat labeled diagram and add a note on accessory structure of the skin.
 - ii) Give details overview of sympathetic and parasympathetic nervous system.
 - iii) Discuss in detail anatomy and physiology of heart. Add a note on blood flow sequence through the heart.
- Q. 3. Short Answers (Answer 7 out of 9) (7 x 5) = 35**
- i) Describe the mechanism of blood coagulation.
 - ii) Describe in detail ribosomes and endoplasmic reticulum.
 - iii) Explain in detail passive transport of material across the cell membrane.
 - iv) Elaborate muscular tissue and connective tissue.
 - v) Explain anatomy and physiology of Eye .
 - vi) Write a note on physiology of muscle contraction.
 - vii) Explain in detail synovial joints.
 - viii) Write a short note on Hematopoiesis.
 - ix) Describe in brief lymph node and spleen.

-----END OF THE PAPER-----

DR. BABASAHEB AMBEDKAR TECHNOLOGICAL UNIVERSITY, LONERE

End Semester Examination – Winter 2022

Date: 10/05/2023

Course : B. Pharmacy
Subject Name : Pharmaceutical Analysis-I
Max Marks : 75

Sem : I
Subject Code : BP102T
Duration : 3 Hr

Instructions:

1. All questions are compulsory
2. Draw diagrams / figures wherever necessary
3. Figures to right indicate full marks

Q. 1. Objective Type Questions (Answer all the questions) (10 x 2) = 20

- i) What is oxidizing agent?
- ii) What base according to Lowery-Bronsted theory?
- iii) Why potentiometer is also called as pH meter?
- iv) Enlist names of limit tests official in Indian Pharmacopeia.
- v) What is mean by primary standard?
- vi) Give the example of self-indicator and external indicator.
- vii) What is mean by equivalent conductance?
- viii) What is masking agent?
- ix) Define Molarity.
- x) What is diazotization titration?

Q. 2. Long Answers (Answer 2 out of 3) (2 x 10) = 20

- i) What is Gravimetry? Explain the steps involved in Gravimetric Analysis.
- ii) Describe the concept conductometric titrations. Explain in detail different types of conductometric titrations along with one example.
- iii) What is mean by redox titration? Enlist its types and types of indicators used in redox titration. Write in detail about Iodine titrations.

Q. 3. Short Answers (Answer 7 out of 9) (7 x 5) = 35

- i) What is impurity? Explain sources of impurities in medicinal agents.
- ii) Differentiate between Mohr's and Volhard's method.
- iii) What is error? Explain in brief types of errors.
- iv) What is potentiometry? Explain types of electrodes used in it along with one example.
- v) What are titration curves? Write in brief various acid-base titration curves.
- vi) Write the principle involved in polarography. Add a note on DME.
- vii) Define Indicator. Add a note on theories of indicators.
- viii) What is non-aqueous titration? Write a note on properties and classes of solvents used in it with example.
- ix) What is complexometric titration? Explain different classes of complexometric titration along with one example.

End Semester Examination – Winter 2022

Date: 06/05/2023

Course : B. Pharmacy
Subject Name : Pharmaceutics-I
Max Marks : 75

Sem: I
Subject Code : BP103T
Duration : 3 Hr.

Instructions:

1. All questions are compulsory
2. Draw diagrams / figures wherever necessary
3. Figures to right indicate full marks

Q. 1. Objective Type Questions (Answer all the questions) (10 x 2) = 20

- i) Enlist the ideal characteristics of ointment base.
- ii) Define pharmaceutical powder? Enlist advantages of powder.
- iii) Differentiate between flocculated & deflocculated suspension
- iv) Find the strength of 75% v/v & 45% v/v alcohol in terms of proof spirit?
- v) Define suspension and enlist the ideal characteristics of suspension.
- vi) What is eutectic mixture? Give one example of it.
- vii) Define syrup and classify it.
- viii) Classify semisolid dosage form along with example.
- ix) Give advantages and disadvantages of liquid dosage form.
- x) Define – a) Liniment, b) Lotion

Q. 2. Long Answers (Answer 2 out of 3) (2 x 10) = 20

- i) Define incompatibility? Enlist different types of it. Explain in detail about physical incompatibility with remedies.
- ii) Define Emulsion? Explain in detail method of preparation of emulsion.
- iii) Define Suppository. Write advantages and disadvantages of suppository. Explain in details method of preparation of suppositories along with evaluation test of suppository.

Q. 3. Short Answers (Answer 7 out of 9) (7 x 5) = 35

- i) Define Prescription and explain the various parts of prescription in detail.
- ii) Write in brief about various identification tests for emulsion.
- iii) Comment on bases used for ointment preparation.
- iv) Define & classify dosage form with suitable examples.
- v) Write a short note on Throat Paint.
- vi) If adult dose of drug is 400mg, what will be the dose of the drug for:
- i) A child of 10 years & ii) 6 months old infant.
- vii) Define suspension. Write stability problems of suspension with remedies to overcome it.
- viii) Give various types of chemical incompatibilities in detail
- ix) Write short note on dusting powder.

-----END OF THE PAPER-----

DR. BABASAHEB AMBEDKAR TECHNOLOGICAL UNIVERSITY, LONERE

End Semester Examination – Winter 2022

Date: 08/05/2023

Course : B. Pharmacy
Subject Name : Pharmaceutical Inorganic Chemistry
Max Marks : 75

Sem : I
Subject Code : BP104T
Duration : 3 Hr.

Instructions:

- All questions are compulsory
- Draw diagrams / figures wherever necessary
- Figures to right indicate full marks

Q. 1. Objective Type Questions (Answer all the questions)

(10 x 2) = 20

- Define Emetics? Give two examples.
- Write down the properties and uses of Boric acid.
- Write down the properties and uses of Ammonium Chloride.
- Define the terms. i) Acidic Buffer ii) Buffer capacity.
- What are Radiopharmaceuticals?
- Define Expectorant. Give two examples.
- Draw a well labeled diagram of Gutzeit apparatus.
- Define Cathartics with examples.
- Write down reaction involved in limit test for Chloride.
- Classify Gastrointestinal agents with examples.

Q. 2. Long Answers (Answer 2 out of 3)

(2 x 10) = 20

- What are Antimicrobial agents? Explain the mechanism of actions of antimicrobial agents with suitable examples. Add a note on Hydrogen Peroxide.
- What is impurity? Enlist and explain sources of impurities in the pharmaceutical substances with examples.
- Write down the role of major Extra and Intra cellular electrolytes with suitable examples. Give the composition of ORS.

Q. 3. Short Answers (Answer 7 out of 9)

(7 x 5) = 35

- Write down the principle, reaction and procedure involved in the limit test of Iron.